

1. It is useful to normalize the wave functions used in quantum mechanics. Why? [1]
2. Can the following wave functions be normalized? If so, calculate the normalization factor.
 - a) $\psi(x) = \sin(\pi x/L)$ on the interval $-\infty < x < +\infty$
 - [4] b) $\psi(x) = \sin(\pi x/L)$ on the interval $0 < x < L$
 - c) $\psi(x) = e^{-i\theta}$ on the interval $0 < \theta < 2\pi$

Hint: A table of integrals might be useful to answer this question. See <http://integral-table.com/>

3. Solving Schrodinger's equation $\hat{H}\psi_n(x) = E_n\psi(x)$ for a particle in a box gives the wave functions

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)$$

The particle is clearly *moving* back and forth in the box, but $\psi_n(x)$ is a **stationary wave** that is *not moving*! Are we missing something here? *Yes!*

Use the **time-dependent Schrodinger equation**

$$\hat{H}\Psi_n(x,t) = i\hbar \left(\frac{\partial \Psi(x,t)}{\partial t} \right)_x$$

- [3] to show the complete wave function for the particle in a box is

$$\Psi_n(x,t) = \psi_n(x)e^{-iE_n t/\hbar}$$

Hints: Because the Hamiltonian operator is a function of x , but not t , $\Psi_n(x,t)$ can be expressed as the product $\psi_n(x)f(t)$ of the function $\psi_n(x)$ that depends only on x and the function $f(t)$ that depends only on t . Use the method of **separation of variables** (recall Assignment #3) to show

$$f(t) = e^{-iE_n t/\hbar}$$

4. For a particle in a box of width L :

$$\langle x \rangle = \frac{L}{2} \quad \langle x^2 \rangle = L^2 \left(\frac{1}{3} - \frac{1}{2\pi^2 n^2} \right) \quad \langle p_x \rangle = 0 \quad \langle p_x^2 \rangle = \frac{n^2 \hbar^2}{4L^2}$$

- [3] Prove the particle obeys the **Heisenberg Uncertainty Principle** $\sigma_x \sigma_{p_x} \geq \hbar/4\pi$.

5. This question refers to a classical particle in a box extending from $x = 0$ to $x = L$.

a) A classical particle is equally likely to be anywhere in the box, so the probability distribution function $P(x)$ is a constant. Use the normalization requirement

$$1 = \int_0^L P(x) dx$$

to show $P(x) = 1/L$.

b) Use $P(x)$ to derive expressions for $\langle x \rangle$, $\langle x^2 \rangle$ and σ_x for a classical particle.

[6] c) Show σ_x for a classical particle in a box equals σ_x for a quantum particle in a box (see Question 4) in the limit $n \rightarrow \infty$.

d) Part c illustrates an important principle of quantum mechanics. Explain.

6. a) An electron with kinetic energy 9.50 eV hits a 10.0 eV potential energy barrier. Calculate the penetration depth.

[2] b) The penetration depth is nonclassical. Why?

7. a) An electron with kinetic energy 11.0 eV hits a 10.0 eV potential energy barrier. Calculate the probability the electron will be reflected by the barrier.

[2] b) The reflection of the electron is nonclassical. Why?

(Q1) Quantum mechanical wave functions are normalized

[for example: $\int \psi^*(x) \psi(x) dx = 1$]

so that $\psi^* \psi$ is a probability distribution

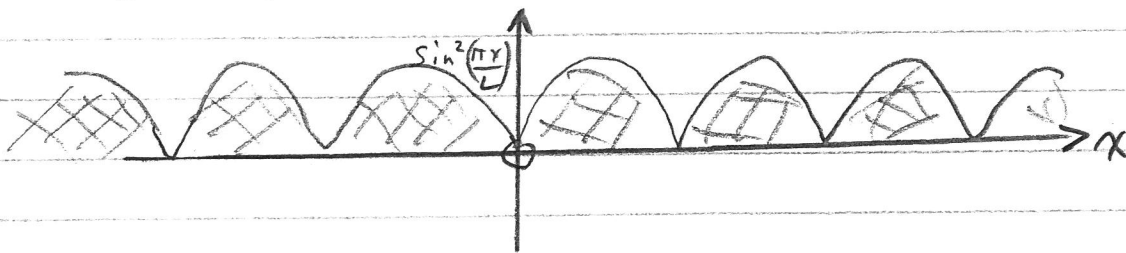
suitable for calculating average $\langle x \rangle, \langle x^2 \rangle, \langle p_x \rangle, \langle E \rangle, \dots$
values

[for example: $\langle x \rangle = \int \psi^*(x) x \psi(x) dx$]

(Q2) a) can $\psi(x) = \sin(\pi x/L)$ be normalized on the interval $-\infty < x < \infty$?

Nope! $\int_{-\infty}^{\infty} \psi^*(x) \psi(x) dx = \int_{-\infty}^{\infty} \sin^2(\pi x/L) dx = \infty$

a graph of $\sin^2(\pi x/L)$:



b) can $\psi(x) = \sin(\pi x/L)$ be normalized on the interval $0 < x < L$?

yes!

(Q2 b cont.) (look up: $\int \sin^2(ax) dx = \frac{x}{2} - \frac{\sin(2ax)}{4a}$)

$$\int_0^L \psi^*(x) \psi(x) dx = \int_0^L \sin\left(\frac{\pi x}{L}\right) \sin\left(\frac{\pi x}{L}\right) dx$$

$$= \int_0^L \sin^2\left(\frac{\pi x}{L}\right) dx = \left(\frac{x}{2} - \frac{\sin(2\pi x/L)}{4\pi/L} \right) \Big|_0^L$$

$$= \left(\frac{x}{2} - \frac{\sin(2\pi)}{4\pi/L} \right) \Big|_0^L = \frac{L}{2} - 0 = \frac{L}{2}$$

To give a normalized wave function, use

$$\psi(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{\pi x}{L}\right)$$

(normalized)

with normalization factor $\sqrt{\frac{2}{L}}$, then

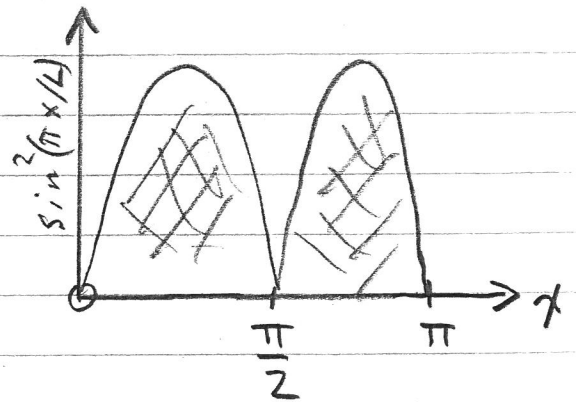
$$\int_0^L \psi^*(x) \psi(x) dx = \int_0^L \sqrt{\frac{2}{L}} \sin\left(\frac{\pi x}{L}\right) \sqrt{\frac{2}{L}} \sin\left(\frac{\pi x}{L}\right) dx$$

normalized

$$= \frac{2}{L} \int_0^L \sin^2\left(\frac{\pi x}{L}\right) dx$$

$$= \frac{2}{L} \frac{L}{2}$$

$$= 1$$



(Q2 cont.)

c) can $\psi(\theta) = e^{-i\theta}$ be normalized on the interval $0 < \theta < 2\pi$? Yes!

$$\int_0^{2\pi} \psi^*(\theta) \psi(\theta) d\theta = \int_0^{2\pi} e^{i\theta} e^{-i\theta} d\theta = \int_0^{2\pi} e^{i\theta - i\theta} d\theta$$

$$= \int_0^{2\pi} e^0 d\theta = \int_0^{2\pi} 1 d\theta = 2\pi$$

the normalization factor here is $\sqrt{\frac{1}{2\pi}}$

$$\psi(\theta)_{\text{normalized}} = \sqrt{\frac{1}{2\pi}} e^{-i\theta}$$

(Q3) time-dependent Schrodinger equation:

$$\hat{H} \Psi(x,t) = i\hbar \left(\frac{\partial \Psi(x,t)}{\partial t} \right)_x$$

using separation of variables: $\Psi(x,t) = \psi(x) f(t)$ and

$$\hat{H}[\psi(x)f(t)] = i\hbar \left(\frac{\partial \psi(x)f(t)}{\partial t} \right)_x$$

$$f(t) \hat{H} \psi(x) = f(t) E \psi(x) = i\hbar \psi(x) \left(\frac{df(t)}{dt} \right)_x$$

$$E f(t) = i\hbar \frac{df(t)}{dt}$$

$$\frac{E}{i\hbar} = \frac{1}{f(t)} \frac{df(t)}{dt} = \frac{d \ln f(t)}{dt} \Rightarrow f(t) = e^{Et/i\hbar}$$

$$f(t) = e^{iEt/i\hbar} = e^{-iEt/\hbar}$$

Q4 $\sigma_x^2 = \langle x^2 \rangle - \langle x \rangle^2$ (the variance of x)

$$= L^2 \left(\frac{1}{3} - \frac{1}{2\pi n^2} \right) - \left(\frac{L}{2} \right)^2$$

$$= L^2 \left(\frac{1}{3} - \frac{1}{2\pi n^2} - \frac{1}{4} \right)$$

$$\left(\begin{aligned} \frac{1}{3} - \frac{1}{4} &= \frac{4}{12} - \frac{3}{12} \\ &= \frac{1}{12} \end{aligned} \right)$$

$$\sigma_x^2 = L^2 \left(\frac{1}{12} - \frac{1}{2\pi n^2} \right)$$

$$\sigma_{p_x}^2 = \langle p_x^2 \rangle - \langle p_x \rangle^2 = \frac{n^2 h^2}{4L^2} - 0$$

$$\sigma_{p_x}^2 = \frac{n^2 h^2}{4L^2}$$

$$\sigma_x \sigma_{p_x} = \sqrt{L^2 \left(\frac{1}{12} - \frac{1}{2\pi n^2} \right)} \sqrt{\frac{n^2 h^2}{4L^2}}$$

$$= \sqrt{\frac{1}{12} - \frac{1}{2\pi n^2}} \frac{nh}{2} \quad \text{(multiply and divide by } 2\pi \text{)}$$

$$= \sqrt{\frac{4\pi^2}{12} - \frac{4\pi^2}{2\pi n^2}} \frac{nh}{2} \frac{1}{2\pi}$$

$$= \sqrt{\frac{\pi^2}{3} - \frac{2}{n^2}} \frac{nh}{2} \quad \text{minimum value for } n=1$$

$$= \sqrt{\frac{\pi^2}{3} - 2} \frac{h}{2} \quad \text{for } n=1$$

$$\min. \sigma_x \sigma_{p_x} = 1.35 \frac{h}{2} > \frac{h}{2}$$

Q5 a) $P(x) = k$ (a constant)

$$1 = \int_0^L P(x) dx = \int_0^L k dx = k \int_0^L dx = kL$$

$$k = \frac{1}{L}$$

$$P(x) = \frac{1}{L}$$

b) average $\langle x \rangle$
x value $\langle x \rangle = \int_0^L x P(x) dx$

$$\langle x \rangle = \int_0^L x \frac{1}{L} dx = \frac{1}{L} \int_0^L x dx = \frac{1}{L} \left. \frac{x^2}{2} \right|_0^L$$

$$= \frac{1}{L} \left(\frac{L^2}{2} - 0 \right) = \frac{L}{2}$$

average $\langle x^2 \rangle$
 x^2 value $\langle x^2 \rangle = \int_0^L x^2 P(x) dx = \frac{1}{L} \int_0^L x^2 dx = \frac{1}{L} \left. \frac{x^3}{3} \right|_0^L$

$$= \frac{1}{L} \left(\frac{L^3}{3} - 0 \right) = \frac{L^2}{3}$$

Variance
of x

$$\sigma_x^2 = \langle x^2 \rangle - \langle x \rangle^2 = \frac{L^2}{3} - \left(\frac{L}{2} \right)^2$$

$$= \frac{L^2}{3} - \frac{L^2}{4} = \frac{L^2}{12}$$

Standard
deviation
of x

$$= \sqrt{\sigma_x^2} = \sigma_x = \frac{L}{\sqrt{12}}$$

(Q5 cont.)

c) for a quantum mechanical particle
in the box

$$\text{variance } \sigma_x^2 \text{ (quantum)} = L^2 \left[\frac{1}{12} - \frac{1}{2\pi n^2} \right]$$

$$\text{standard deviation } \sigma_x \text{ (quantum)} = L \sqrt{\frac{1}{12} - \frac{1}{2\pi n^2}}$$

as $n \rightarrow \infty$ (very large quantum numbers)

notice

$$\sigma_x \text{ (quantum)} \lim_{n \rightarrow \infty} = L \sqrt{\frac{1}{12}}$$

$$= \sigma_x \text{ (classical)} = \frac{L}{\sqrt{12}}$$

d) part c illustrates the Correspondence Principle:

as quantum numbers get very large, many energy levels are populated, and the discrete quantum energy levels become continuous (classical)

Q6 a) a 9.50 eV electron hits a 10.0 eV barrier.

$$\begin{aligned} \text{penetration depth } D_p &= \frac{h/2\pi}{\sqrt{(V_0 - E)2m}} \\ &= \frac{6.626 \times 10^{-34} / 2\pi}{\sqrt{(10.0 - 9.50)(1.602 \times 10^{-19})2} \cdot 9.110 \times 10^{-31}} \\ &= 2.76 \times 10^{-10} \text{ m} \quad \boxed{0.276 \text{ nm}} \end{aligned}$$

b) classically, the electron has insufficient kinetic energy to penetrate the barrier, and $D_p = 0$

Q7 an electron with 11.0 eV hits a 10.0 eV barrier

a)

$$R = \text{probability of reflection} = \frac{2y - 2\sqrt{y(y-1)} - 1}{2y + 2\sqrt{y(y-1)} - 1} \quad \left(\begin{array}{l} y = \frac{E}{V_0} \\ = 1.10 \end{array} \right)$$

$$R = \frac{2(1.10) - 2\sqrt{1.10(1.10)} - 1}{2(1.10) + 2\sqrt{1.10(1.10)} - 1} = \frac{0.536}{1.863}$$

$$\boxed{R = 0.288}$$

b) classically, for $E > V_0$, the probability of reflection is zero